### **Eigenvalues of the Angular Momentum Operators**

Today, we are talking about the eigenvalues of the angular momentum operators. **J** is used to denote angular momentum in general, **L** is used specifically to denote orbital angular momentum, and **S** is used to denote spin. J = L + S [Note: follows Section 9.2, page 358]

J is generalized; we still have the commuter relations:

$$\begin{bmatrix} \hat{J}_{x}, \hat{J}_{y} \end{bmatrix} = i\hbar \hat{J}_{z} 
\begin{bmatrix} \hat{J}_{y}, \hat{J}_{z} \end{bmatrix} = i\hbar \hat{J}_{x} 
\begin{bmatrix} \hat{J}_{z}, \hat{J}_{x} \end{bmatrix} = i\hbar \hat{J}_{y}$$

$$\vec{J} = \vec{L} + \vec{S} = \text{total moment}$$

Today, we want to find the eigenvalues of  $J_z$  and  $J^2$ . The generalized relation is applicable to the z axis. We will show that:

$$\hat{J}_z = m\hbar$$
 where  $m = -j, -j + 1, ..., 0, j - 1, j$   
 $\hat{J}^2 = \hbar j (j + 1)$  where  $j = \frac{n}{2}$  and  $n = 0, 1, 2, ...$ 

Compare this to the orbital angular momentum operators:

$$\hat{L}_z = m\hbar$$
 where  $m = -j, -j + 1, ..., 0, j - 1, j$   
 $\hat{L}^2 = \hbar l(l+1)$  where  $l = 0, 1, 2, ...$ 

# **Ladder Operators**

From the commutator relations and from the creation/annihilation operators from the Harmonic oscillator, we have the Ladder Operators for angular momentum.

$$\hat{J}_{+} = \hat{J}_{x} + i\hat{J}_{y}$$

$$\hat{J}_{-} = \hat{J}_{x} - i\hat{J}_{y}$$

 $\hat{J}^{2}$  commutes with the otehr components of angular momentum

$$\left[\hat{J}_{x},\hat{J}^{2}\right] = \left[\hat{J}_{y},\hat{J}^{2}\right] = \left[\hat{J}_{z},\hat{J}^{2}\right] = 0$$

Some other properties of these operators:

$$\begin{bmatrix}
\hat{J}_z, \hat{J}_+ \\
\hat{J}_z, \hat{J}_-
\end{bmatrix} = \hbar \hat{J}_+$$
We want to prove this directly.

Here is our proof:

We show this explicitly:

$$\begin{split} \left[\hat{J}_{z},\hat{J}_{+}\right] &= \left[\hat{J}_{z},\hat{J}_{x} + i\hat{J}_{y}\right] \\ &= \hat{J}_{z}\left(\hat{J}_{x} + i\hat{J}_{y}\right) - \left(\hat{J}_{x} + i\hat{J}_{y}\right)\hat{J}_{z} \\ &= \hat{J}_{z}\hat{J}_{x} + i\hat{J}_{z}\hat{J}_{y} - \hat{J}_{x}\hat{J}_{z} - i\hat{J}_{x}\hat{J}_{y} \\ &= \left[\hat{J}_{z},\hat{J}_{x}\right] + i\left[\hat{J}_{z},\hat{J}_{y}\right] \\ &= i\hbar\hat{J}_{y} - i^{2}\hbar\hat{J}_{x} \\ &= \hbar\left(\hat{J}_{x} + i\hat{J}_{y}\right) \\ &= \hbar\hat{J}_{+} \end{split}$$

We can change the order to have a new relation:

$$\hat{J}_{z}\hat{J}_{+}=\hat{J}_{+}\hat{J}_{z}+\hbar\hat{J}_{+}$$

Show that  $\hat{J}^2 = \hat{J}_{\pm}\hat{J}_{\pm} + \hat{J}_z^2 \pm \hbar\hat{J}_z$ 

We want to prove this relation directly.

$$\begin{split} \hat{J}_{-}\hat{J}_{+} &= \left(\hat{J}_{x} - i\hat{J}_{y}\right) \left(\hat{J}_{x} + i\hat{J}_{y}\right) \\ &= \hat{J}_{x}\hat{J}_{x} + \hat{J}_{x}i\hat{J}_{y} - i\hat{J}_{y}\hat{J}_{x} - i\hat{J}_{y}i\hat{J}_{y} \\ &= \hat{J}_{x}^{2} + i\hat{J}_{x}\hat{J}_{y} - i\hat{J}_{y}\hat{J}_{x} + \hat{J}_{y}^{2} \\ &= \hat{J}_{x}^{2} + \hat{J}_{y}^{2} + i[\hat{J}_{x}, \hat{J}_{y}] \\ &= \hat{J}_{x}^{2} + \hat{J}_{y}^{2} - \hbar\hat{J}_{z} \\ &\therefore \hat{J}_{x}^{2} + \hat{J}_{y}^{2} + \hat{J}_{z}^{2} = \hat{J}_{-}\hat{J}_{+} + \hat{J}_{z}^{2} + \hbar\hat{J}_{z} \\ \hat{J}^{2} &= \hat{J}_{-}\hat{J}_{+} + \hat{J}_{z}^{2} + \hbar\hat{J}_{z} \end{split}$$

#### **Prove the Commutator Relations**

We can use these relations – put directly – to find the commutator relations.

$$\hat{J}^{2} = \hat{J}_{x}^{2} + \hat{J}_{y}^{2} + \hat{J}_{z}^{2}$$

$$\hat{J}^{2} = \hat{J}_{\pm}\hat{J}_{\pm} + \hat{J}_{z}^{2} \pm \hbar\hat{J}_{z}$$

$$\begin{split} \left[\hat{J}_{+},\hat{J}_{-}\right] &= 2\hbar\hat{J}_{z} \\ \left[\hat{J}_{+},\hat{J}_{-}\right] &= \hat{J}_{+}\hat{J}_{-} - \hat{J}_{-}\hat{J}_{+} \\ &= \left(\hat{J}_{x} + i\hat{J}_{y}\right)\left(\hat{J}_{x} - i\hat{J}_{y}\right) - \left(\hat{J}_{x} - i\hat{J}_{y}\right)\left(\hat{J}_{x} + i\hat{J}_{y}\right) \\ &= \hat{J}_{x}\hat{J}_{x} - \hat{J}_{x}i\hat{J}_{y} + i\hat{J}_{y}\hat{J}_{x} - i\hat{J}_{y}i\hat{J}_{y} - \hat{J}_{x}\hat{J}_{x} - \hat{J}_{x}i\hat{J}_{y} + i\hat{J}_{y}\hat{J}_{x} + i\hat{J}_{y}i\hat{J}_{y} \\ &= 2i\left(\hat{J}_{y}\hat{J}_{x} - \hat{J}_{x}\hat{J}_{y}\right) \\ &= 2i\left[\hat{J}_{y},\hat{J}_{x}\right] \\ &\therefore \left[\hat{J}_{y},\hat{J}_{x}\right] = -i\hbar\hat{J}_{z} \\ &\left[\hat{J}_{x},\hat{J}_{y}\right] = i\hbar\hat{J}_{z} \end{split}$$

Also we can show:

$$\hat{J}^{2} = \hat{J}_{-}\hat{J}_{+} + \hat{J}_{z}^{2} + \hbar\hat{J}_{z}$$

$$+\hat{J}^{2} = \hat{J}_{+}\hat{J}_{-} + \hat{J}_{z}^{2} - \hbar\hat{J}_{z}$$

$$2\hat{J}^{2} = \hat{J}_{-}\hat{J}_{+} + \hat{J}_{z}^{2} + \hbar\hat{J}_{z} + \hat{J}_{+}\hat{J}_{-} + \hat{J}_{z}^{2} - \hbar\hat{J}_{z}$$

$$\therefore \left[\hat{J}_{-}\hat{J}_{+} + \hat{J}_{+}\hat{J}_{-} = 2(\hat{J}^{2} - \hat{J}_{z}^{2})\right]$$

# Eigenvalues of $\hat{J}^2$ and $\hat{J}_z$

To solve the problem of  $\hat{J}^2$  and  $\hat{J}_z$  to solve eigenvalue of these two physical parameters – first we see that they commute to each other.

$$\left[\hat{J}^2,\hat{J}_z\right]=0$$

Here is the eigenvalue equation:

$$\hat{m{J}}_zm{arphi}_m=\mathop{\hbar\!m}\limits_{ ext{eigenvalue}}m{arphi}_m$$

Use the commutator relations to change the order of these operators.

$$\begin{split} \left[\hat{J}_{z},\hat{J}_{+}\right] &= \hat{J}_{z}\hat{J}_{+} - \hat{J}_{+}\hat{J}_{z} = \hbar\hat{J}_{+} \Rightarrow \hat{J}_{z}\hat{J}_{+} = \hat{J}_{+}\hat{J}_{z} + \hbar\hat{J}_{+} \\ \hat{J}_{z}\hat{J}_{+}\varphi_{m} &= \left(\hat{J}_{+}\hat{J}_{z} + \hbar\hat{J}_{+}\right)\varphi_{m} = \hat{J}_{+}\hat{J}_{z}\varphi_{m} + \hbar\hat{J}_{+}\varphi_{m} = \hat{J}_{+}m\hbar\varphi_{m} + \hbar\hat{J}_{+}\varphi_{m} = \hbar(m+1)\hat{J}_{+}\varphi_{m} \\ \hat{J}_{z}\left(\hat{J}_{+}\varphi_{m}\right) &= \hbar(m+1)\left(\hat{J}_{+}\varphi_{m}\right) \end{split}$$

So this implies that:

$$\hat{J}_{+}\varphi_{m} = \varphi_{m+1}$$

$$\hat{J}_{z}\hat{J}_{+}\varphi_{m} = \hbar(m+1)\varphi_{m+1}$$

Similarly:

$$\begin{split} & \left[ \hat{J}_{z}, \hat{J}_{-} \right] = \hat{J}_{z} \hat{J}_{-} - \hat{J}_{-} \hat{J}_{z} = -\hbar \hat{J}_{-} \Rightarrow \hat{J}_{z} \hat{J}_{-} = \hat{J}_{-} \hat{J}_{z} - \hbar \hat{J}_{-} \\ & \hat{J}_{z} \hat{J}_{-} \varphi_{m} = \left( \hat{J}_{-} \hat{J}_{z} - \hbar \hat{J}_{-} \right) \varphi_{m} = \hat{J}_{-} \hat{J}_{z} \varphi_{m} - \hbar \hat{J}_{-} \varphi_{m} = \hat{J}_{-} m \hbar \varphi_{m} - \hbar \hat{J}_{-} \varphi_{m} = \hbar (m-1) \hat{J}_{-} \varphi_{m} \\ & \hat{J}_{z} \left( \hat{J}_{-} \varphi_{m} \right) = \hbar (m-1) \left( \hat{J}_{-} \varphi_{m} \right) \end{split}$$

So this implies that:

$$\hat{J}_{-}\varphi_{m} = \varphi_{m-1}$$

$$\hat{J}_{z}\hat{J}_{-}\varphi_{m} = \hbar(m-1)\varphi_{m-1}$$

Now we need to find the common eigenfunctions – the eigenfunction  $\varphi_{\scriptscriptstyle m}$  of  $\hat{J}_{\scriptscriptstyle z}$  and the eigenfunction  $\varphi_{\scriptscriptstyle k}$  of  $\hat{J}^{\scriptscriptstyle 2}$ . Common functions found from:

$$\hat{J}^2 \varphi_m = \hbar^2 k^2 \varphi_k$$
 where  $k$  is any real integer

We calculate the expectation value:

$$\langle J^{2} \rangle = \int \varphi_{m} \hat{J}^{2} \varphi_{m} dx = \hbar^{2} k^{2}$$
also:
$$\langle J^{2} \rangle = \langle J_{x}^{2} \rangle + \langle J_{y}^{2} \rangle + \langle J_{z}^{2} \rangle$$

$$\geq 0 \qquad \geq 0 \qquad = \hbar^{2} m^{2}$$

$$\therefore \hbar^{2} k^{2} \geq \hbar^{2} m^{2}$$

$$k^{2} \geq m^{2} \quad \text{if} \quad k > 0 \quad \text{then} \quad -k \leq m \leq k$$
we have this confinement

m has a maximum and also a minimum

$$\hat{J}_{+}\varphi_{m\max} = 0$$
 and  $\hat{J}_{-}\varphi_{m\min} = 0$   
 $\hat{J}^{2}\varphi_{m\max} = \hbar^{2}k^{2}\varphi_{m\max}$  and  $\hat{J}^{2}\varphi_{m\min} = \hbar^{2}k^{2}\varphi_{m\min}$ 

To find the eigenvalues of  $\hat{J}^2$  and  $\hat{J}_z$  we use the commutator relations  $\hat{J}^2 = \hat{J}_{\pm}\hat{J}_{\pm} + \hat{J}_z^2 \pm \hbar\hat{J}_z$ :

$$\begin{split} \hat{J}^2 \varphi_{m\text{max}} &= \left(\hat{J}_- \hat{J}_+ + \hat{J}_z^2 + \hbar \hat{J}_z\right) \varphi_{m\text{max}} = \left(0 + \hbar^2 m_{\text{max}}^2 + \hbar^2 m_{\text{max}}\right) \varphi_{m\text{max}} = \hbar^2 m_{\text{max}} \left(m_{\text{max}} + 1\right) \varphi_{m\text{max}} \\ \hat{J}^2 \varphi_{m\text{max}} &= \hbar^2 k^2 \varphi_{m\text{max}} \\ \hbar^2 k^2 \varphi_{m\text{max}} &= \hbar^2 m_{\text{max}} \left(m_{\text{max}} + 1\right) \\ k^2 &= \hbar^2 m_{\text{max}} \left(m_{\text{max}} + 1\right) \end{split}$$

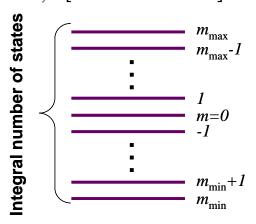
Look at this using the same method to do minimum.

$$\begin{split} \hat{J}^2 \varphi_{m \min} &= \left(\hat{J}_+ \hat{J}_- + \hat{J}_z^2 - \hbar \hat{J}_z\right) \varphi_{m \min} = \left(0 + \hbar^2 m_{\min}^2 - \hbar^2 m_{\min}\right) \varphi_{m \min} = \hbar^2 m_{\min} \left(m_{\min} - 1\right) \varphi_{m \min} \\ \hat{J}^2 \varphi_{m \min} &= \hbar^2 k^2 \varphi_{m \min} \\ \hbar^2 k^2 \varphi_{m \min} &= \hbar^2 m_{\min} \left(m_{\min} - 1\right) \varphi_{m \min} \\ k^2 &= \hbar^2 m_{\min} \left(m_{\min} - 1\right) \end{split}$$

Let:

$$m_{\text{max}} \equiv j$$

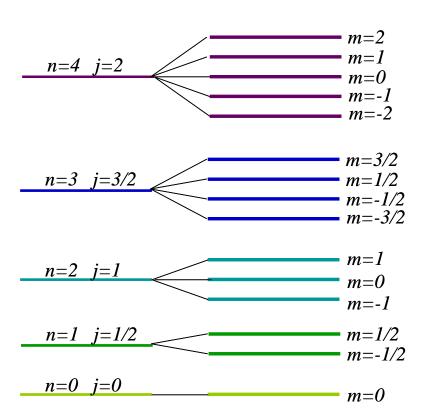
From the eigenfunction you have  $J_z = \hbar m_j$ . From the ladder operator, we have the range of values  $m_j = [-j, -j+1, ..., j-1, j]$ . Here are the values:



Here are the eigenvalues of  $\hat{J}^2$  and  $\hat{J}_z$ . We got this result from the commutator relations  $\hat{J}^2 = \hat{J}_{\scriptscriptstyle \mp} \hat{J}_{\scriptscriptstyle \pm} + \hat{J}_z^2 \pm \hbar \hat{J}_z$ .

$$J^{2} = \hbar^{2} j (j+1) \qquad j = \frac{n}{2} \qquad n = 0,1,2,...$$

$$J_{z} = \hbar m \qquad m = -j, -j+1,...,0,..., j-1, j$$



Similar to figure 9.6 on page 362.

## **Rotational Energy Levels of Molecules**

The rotational level of a molecule is the lowest energy level – is rotation.

From classical mechanics:

$$\hat{H} = \frac{L^2}{2I} \quad \text{and} \quad I = 2a^2 M$$

We are not considering vibration, so a is constant. The eigenvalues are:  $L^2 = \hbar^2 l(l+1)$ 

$$E_{l} = \frac{1}{2I} \hbar^{2} l (l+1) = \frac{1}{4Ma^{2}} \hbar^{2} l (l+1)$$

$$E_{l+1} - E_l = \frac{1}{4Ma^2} \hbar^2 \left[ (l+1)(l+2) - l(l+1) \right] = \frac{1}{4Ma^2} \hbar^2 \left[ (l+1)(l+2-l) \right] = \frac{1}{2Ma^2} \hbar^2 \left[ (l+1$$

Using the same method, same eigenvalue equation – talk about Hydrogen – the simplest molecule M-M and the  $\Delta E = E_l - E_{l-1}$ . Here  $\frac{\hbar^2}{2m_e a_{Bohr}^2} = 13.6 \text{ eV}$  is called the Rydberg constant. It is the energy of ionizing a hydrogen atom.

If we consider this Rydberg constant to be  $\frac{\hbar^2}{2Ma^2}$  is 13.6 eV $\left(\frac{m_e}{M}\right)$  where  $\left(\frac{m_e}{M}\right)$  is usually on the order of  $10^{-3}$ . So the Rydberg constant is about 10meV for Hydrogen (about 1K energy and  $\lambda \rightarrow$ mm. Remember for the qualifying exam – if you know this -  $10 \cdot 10^{-3}$  eV.

Homework: 9.5 and 9.6 – not difficult – simple like general physics.