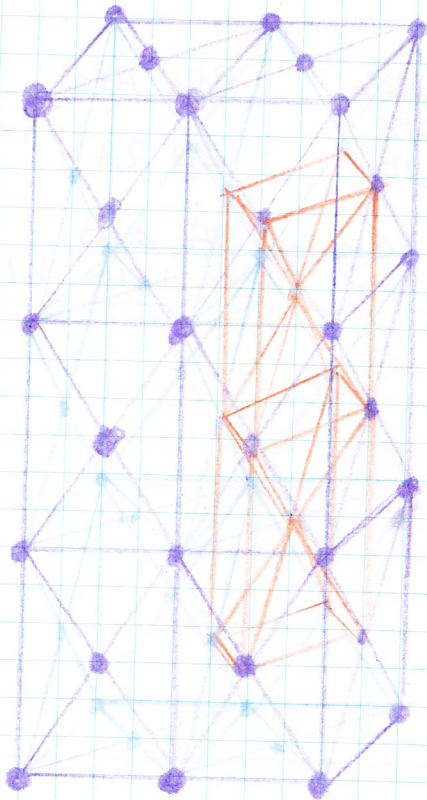


1) In the tetragonal crystal system there are two allowed lattice centerings, namely the primitive (P) and the body-centered (I). Consider a face-centered (F) tetragonal lattice.

a) Show that the lattice is equivalent to P or I.

FCC below in purple is equivalent to BCC below in orange.



b) Assuming the FCC lattice has lattice constants \vec{a} and \vec{c} , what are the equivalent in BCC?

$$\vec{c}_{\text{FCC}} = \vec{c}_{\text{BCC}} \quad \frac{\sqrt{2}}{a} \vec{a}_{\text{FCC}} = \vec{a}_{\text{BCC}}$$

c) Why does not the same equivalence hold in cubic crystal systems?

There is no way to define a face centered cubic lattice with the BCC system. The center point of the BCC generates both the 8 points of the simple crystal or primitive component and the center points of the faces in the x-y plane. These are not both integral values of a side of a cube.

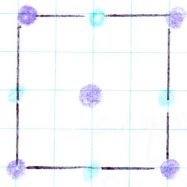
7

2.) Show the representations of the following crystal structures in the cubic system as a set of stacked layers using square lattices a, b, c, d, e, f, g, h .

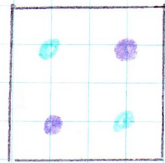
<p>$Pm\bar{3}m$ (#221)</p> <p>αReO_3</p> <p>A B f a A e ✓ $1\alpha Re, 2O$ 1O ✓ 1A, 2B's 1B</p> <p>2 layers 1A, 3B fits αReO_3 ratio 1:3</p>	<p>$Pm\bar{3}m$ (#221)</p> <p>Cu_3Au</p> <p>B e B e a B f a A f a ✓ A e n B e 2A 2A ✓ 1A, 1B 2A</p> <p>2 layers 3A, 2B fits Cu_3Au ratio 3:1</p>
<p>NbO</p> <p>$Pm\bar{3}m$ (#221)</p> <p>A e n B f a A e B e a 2B 1A A f a B e 1B 2A</p> <p>2 layers 3B, 3A fits ratio NbO 1:1</p>	
<p>$Fm\bar{3}m$ (#225)</p> <p>CaF_2 Fluorite</p> <p>A f, 2A's B h d, 4B's A f a, 2A's B h d, 4B's</p> <p>4 layers 4A's 8B's fits 1:2 ratio ✓</p>	9
<p>$Fm\bar{3}m$ (#225)</p> <p>AlF_3</p> <p>A f A B f B h d 2A's, 2B's 4B's</p> <p>A f, B f a B h d 2A's 2B's 4B's</p> <p>4 layers 4A's, 12B's fits 1:3 ratio</p>	9

2.) continued...

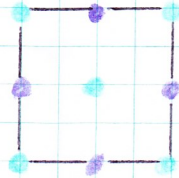
Fd $\bar{3}m$ NaCl (#227)



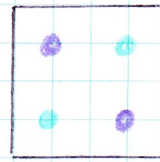
Af Bfa
2A, 2B



Afd, Bfd₂
2A, 2B



Afa Bf
2A, 2B

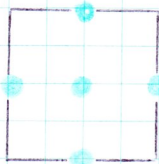


Afd₂ Bfd₁
2A, 2B ✓

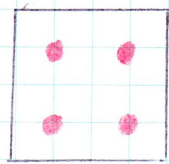
4 layers
8A, 8B

ratio 1:1
works for NaCl

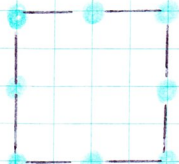
Im $\bar{3}m$ Pt₃O₄
(#229)



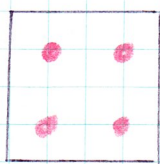
Aga Agn
3A



Bhd
4B



Ag
3A



Bhd
4B

4 layers ✓

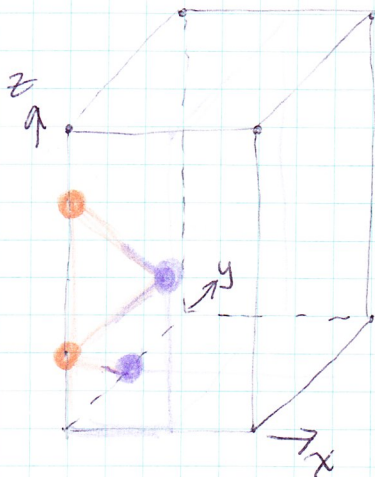
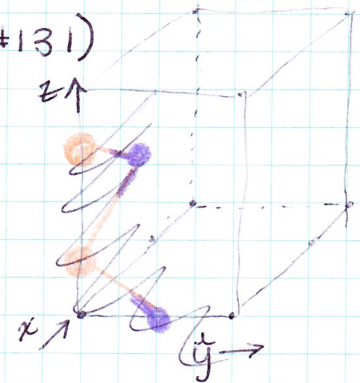
3.) Consider the PES structure (P4₂/mmc) (#131)

Primitive Vectors

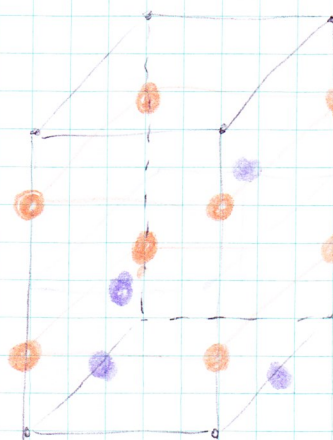
$$\begin{aligned} \vec{A}_1 &= a\hat{x} \\ \vec{A}_2 &= a\hat{y} \\ \vec{A}_3 &= c\hat{z} \end{aligned}$$

Basis Vector

$$\begin{aligned} \vec{B}_1 &= \frac{1}{2}a\hat{y} && \text{Pt} \\ \vec{B}_2 &= \frac{1}{2}a\hat{x} + \frac{1}{2}a\hat{z} && \text{Pt} \\ \vec{B}_3 &= \frac{1}{4}\hat{z} && \text{S} \\ \vec{B}_4 &= \frac{3}{4}\hat{z} && \text{S} \end{aligned}$$

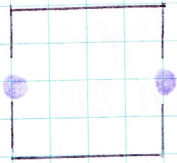


unit cell
2S and 2Pt



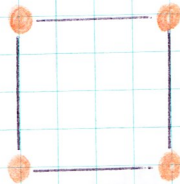
4-fold symmetry

Pt S Stacked Layers



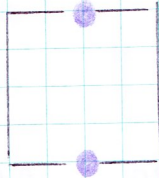
A ~~Be~~ a_2

IA



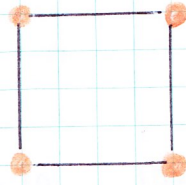
Be

IB



A ~~Be~~ a_1

IA ~~IB~~



Be

IB

4 layers

2A, 2B

ratio 1:1 PLS

Handwritten signature